Chemistry named after. A.P. Arzamastsev Moscow, Russia



Development of the method HPTLC for the quantitative and qualitative determination of phenothiazine derivatives

Podoshvina Anna Andreevna, student Tamara Kobakhidze, assistant

Goal of the reserch.

To improve the existing TLC techniques for the HPTLS method.

Objectives of the experiment.

1) To analyze the existing TLC and HPTLC methods. 2) The choice of the methodology on the basis of which the quantitative and qualitative analysis of phenothiazine derivatives will be developed.

Analysis of TLC and HPTLC methods in



In TLC of phenothiazine derivatives, the following detectors are used: an FPN reagent (a mixture of 5% FeCl3) solution, 20% HClO4 solution and 50% HNO3 solution in a ratio of 5:45:50), a solution of perchloric acid, iodoplatinate, a solution of concentrated sulfuric acid.



Brief description of the danger of perchloric acid: a strong oxidizer; may cause fire or explosion, may cause corrosion of metals, chemical burns if it gets on the skin.

The use of detectors is not necessary in the HPTLC methodology we are developing. Thus, we will be able to reduce the use of reagents that are dangerous both for the environment and for workers.

Materials and methods of research.

solution

Pharmacopoeias, available articles and journals, studies on phenothiazion derivatives,

Main results

The efficiency of the separation of substances largely depends on the choice of the mobile Phenothiazine pKa phase 9,3 Chlorpromazine Based on the basic properties of drugs, the following variants of organic solvent systems can be distinguished as mobile phases for the analysis of Levomepromazine phenothiazine derivatives: · methanol-25% ammonia solution (100:1.5) Diprazine 9,1 · cyclohexane-toluene-diethylamine (75:15:10) acetate-methanol-25% ammonia ethyl (85:105)Thioridazine 9,5 · chloroform-ethanol(90:10) ·acetone-methanol-25%ammonia solution(90:10:2) 9,4 Promazin methanol-n-butanol (60:40)

The Pharmacopoeia of the Eurasian Economic Union - petroleum ether P diethylamine P (50:1, vol/vol) saturated with phenoxyethanol P Japanese Pharmacopoeia Acetone-hexane-ammonia solution(16:6:1) Pharmacopoeia of the Russian Federation - acetone-diethylaminecyclohexane (10:10:80)

benzene-dioxane-ammonia(75:20:5)

· chloroform-methanol(90:10)

The chemical structure of drugs of this group is based on the heterocyclic

system of phenothiazine, which includes nitrogen and sulfur heteroatoms.

1		
Characteristics of the plates	TLC	HPTLC
Average particle size, mkm	11-20	5-7
Layer thickness, mkm	250	100
The length of the solvent front run, cm	10-15	3-5
Amount of solvent, ml	50	5-10
Detection limit, mcg	1	0,1
Spot diameter, mm	2-4	1

8,1

Trifluoperazine

-one-component system - ethyl acetate -single-component eluent - methanol -toluene-ethyl ether-chloroform (30:50:20, vol/vol) -benzene-dioxane-25% ammonia solution (60:35:5) -ethyl acetate-acetone-25% ammonia solution in ethanol 1:1 (50:45:4)

-toluene-acetone-ethanol-25% ammonia

Due to the narrower distribution of silica gel, the spots of the separated substances become more compact and are better separated by a shorter distance in less elution time.

The accuracy of the applied volume and the compactness of the zone subsequently affect the final result of quantitative determination. The quantitative assessment of the spot in the device takes place according to its two characteristics: by the area of the spot and its "volume" in space, the intensity of the stain color (brightness) is used as the third coordinate.

The efficiency of the separation of substances largely depends on the choice of the mobile phase.

Conclusions.

Thus, when conducting a literature review on this topic, it is planned to develop a methodology for the qualitative and quantitative determination of compounds derived from phenothiazines by the VETSC method in combination with densitometry for three phenothiazine derivatives in the acetonemethanol-25% ammonia solution system